CH 310N Fall 2006

Anslyn

September 28th, 2006

Exam 1

Print Name	UT-EID	
Half of these problems are from Problem 16.15 from your ho	mework!> 1)	(12 pts)
	2)	(8 pts)
	3)	(8 pts)
Half of these problems are from your homework!	> 4)	(39pts)
	5)	(10 pts)
	6)	(10 pts)
	7)	(10 pts)
Problem 15.11 from your homework!	> 8)	(8 pts)
Problem 16.55 from your homework!		(8 pts)
	T 1	(112

Total score _____ (113pts)

1) Name the structure or draw the structure of the nomenclature provided in the box.

(12 points)

A)

Draw formaldehyde

B) From homework, problem 16.15.

Draw 3-methyl-3-buten-2-one



C) Name the structure



D) From homework, problem 16.15.

Draw 5-oxohexanal

2) The following is the ¹H-NMR spectra for C_4H_6O . Draw the chemical structure of this compound in the box provided and provide the IUPAC name of the structure using the spectrums provided below. (8 points)



Expansion of peaks:







Expansion Pictures:



- 4) Fill in the box with the correct reagent, reactant or product. (39 points)
- A) From homework, problem 15.7





B) From homework, problem 16.30.



D)



E)





I) From homework, problem 16.38.

 CH_2I_2





5) Provide a mechanism for the following reaction. Show all arrow pushing and intermediates. Show any lone pair electrons you use as a nucleophilic source or that are created during arrow pushing. (10 points)

 $\begin{array}{c} 0 \\ H_{3}0, H_{2}0 \\ \end{array} \\ \end{array} \\ \begin{array}{c} 0 \\ H_{3}0, H_{2}0 \\ \end{array} \\ \end{array} \\ \begin{array}{c} 0 \\ H_{3}0, H_{2}0 \\ \end{array} \\ \end{array} \\ \begin{array}{c} 0 \\ H_{3}0, H_{2}0 \\ \end{array} \\ \end{array}$ \\ \begin{array}{c} 0 \\ H_{3}0, H_{2}0 \\ \end{array} \\ \begin{array}{c} 0 \\ H_{3}0, H_{2}0 \\ \end{array} \\ \end{array} \\ \begin{array}{c} 0 \\ H_{3}0, H_{3}0 \\ \end{array} \\ \begin{array}{c} 0 \\ H_{3}0, H_{3}0 \\ \end{array} \\ \end{array} \\ \begin{array}{c} 0 \\ H_{3}0, H_{3}0 \\ \end{array} \\ \end{array} \\ \begin{array}{c} 0 \\ H_{3}0, H_{3}0 \\ \end{array} \\ \end{array} \\ \begin{array}{c} 0 \\ H_{3}0, H_{3}0, H_{3}0 \\ H_{3}0, H_{3}0 \\ \end{array} \\ \end{array} \\ \begin{array}{c} 0 \\ H_{3}0, H_{3}0, H_{3}0 \\ H_{3}0, H_{3}0 \\ H_{3}0, H_{3}0 \\ H_{3}0, H_{3}0, H_{3}0 \\ H_{3}0, H_{3}0, H_{3}0 \\ H_{3}0, H_{3}0, H_{3}0 \\ H_{3}0, H_{3}0

6) Draw the products of the following reactions. Propose a mechanism for the reactions. Be sure to show all intermediates and curved arrows for each step. Show any lone pair electrons you use as a nucleophilic source or that are created during arrow pushing. (10 points)



7) Draw the products of the following reactions. Propose a mechanism for the reactions. Be sure to show all intermediates and curved arrows for each step. Show any lone pair electrons you use as a nucleophilic source or that are created during arrow pushing. (10 points)

NaOH, Br2

8) Show how to synthesize:



Use only the reagents given as a source of carbon. You may use any inorganic reagents necessary. Do not worry about arrow pushing (mechanism). (8 points) (Problem 15.11 in homework)

9) Show a synthesis of
OH from H————H, H, and CH₃I
Use only the reagents given as a source of carbon. You may use any inorganic reagents necessary. Do not worry about arrow pushing (mechanism). (8 points) (Problem 16.55 in homework)