CH 310N Fall 2006

Anslyn

September 28th, 2006

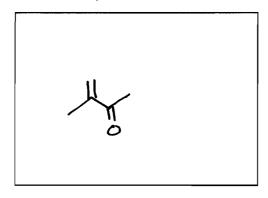
Exam 1

Print Name		UT-EID	
Half of these problems are	from Problem 16.15 from your h	omework!> 1)	(12 pts)
		2)	(8 pts)
		3)	(8 pts)
Half of these problems	s are from your homework!-	> 4)	(39pts)
		5)	(10 pts)
		6)	(10 pts)
		7)	(10 pts)
Problem 15.11	from your homework!	> 8)	(8 pts)
Problem 16.5	55 from your homework!	>9)	(8 pts)
		Total score	(113pts)

- 1) Name the structure or draw the structure of the nomenclature provided in the box. (12 points)
- A) Draw formaldehyde

B) From homework, problem 16.15.

Draw 3-methyl-3-buten-2-one

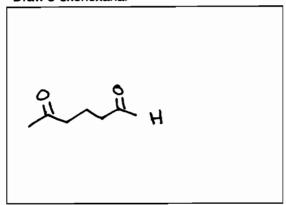


C) Name the structure

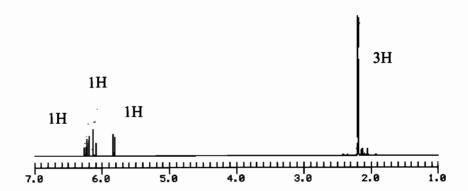
65-Bromo-4-methyl-4E-octen-2-one

D) From homework, problem 16.15.

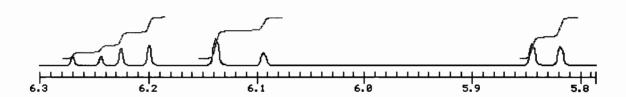
Draw 5-oxohexanal

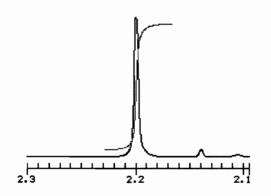


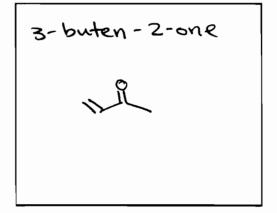
2) The following is the ¹H-NMR spectra for C₄H₆O. Draw the chemical structure of this compound in the box provided and provide the IUPAC name of the structure using the spectrums provided below. (8 points)



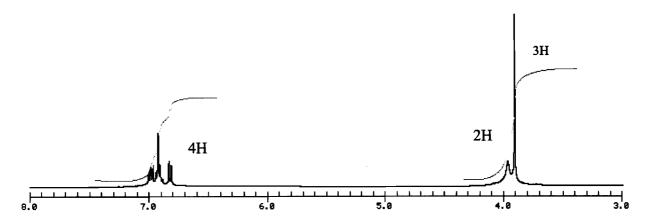
Expansion of peaks:



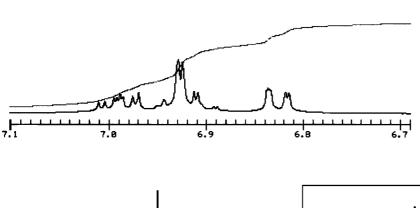


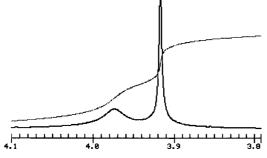


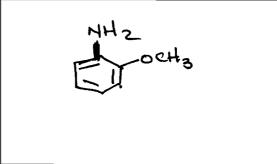
3) Following is the ¹H-NMR spectra for C₇H₉NO. Draw the chemical structure of this compound in the box provided from the spectrums provided. (8 points)



Expansion Pictures:

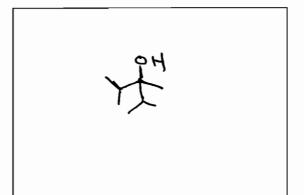




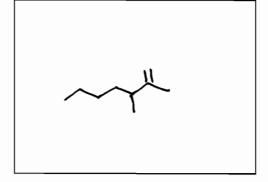


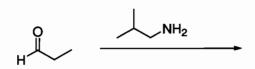
- 4) Fill in the box with the correct reagent, reactant or product. (39 points)
- A) From homework, problem 15.7

B) From homework, problem 16.30.

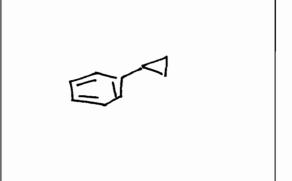


E)



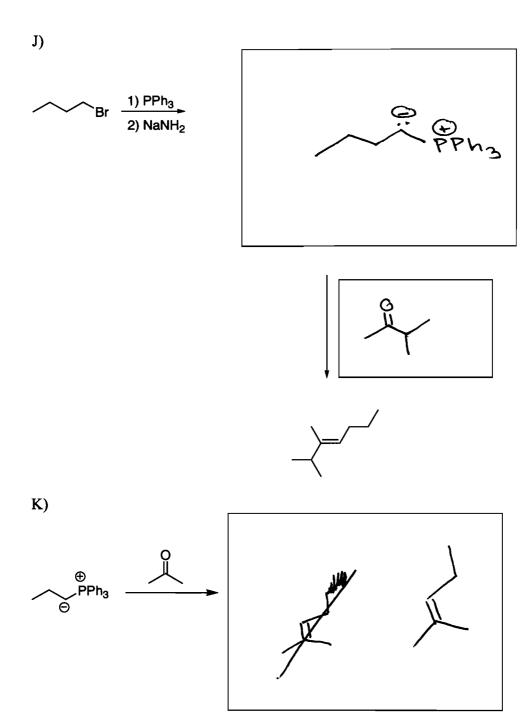


H) From homework, problem 15.12.



I) From homework, problem 16.38.

$$H$$
 H_2NNH_2



5) Provide a mechanism for the following reaction. Show all arrow pushing and intermediates. Show any lone pair electrons you use as a nucleophilic source or that are created during arrow pushing. (10 points)

6) Draw the products of the following reactions. Propose a mechanism for the reactions. Be sure to show all intermediates and curved arrows for each step. Show any lone pair electrons you use as a nucleophilic source or that are created during arrow pushing. (10 points)

7) Draw the products of the following reactions. Propose a mechanism for the reactions. Be sure to show all intermediates and curved arrows for each step. Show any lone pair electrons you use as a nucleophilic source or that are created during arrow pushing. (10 points)

8) Show how to synthesize:

Use only the reagents given as a source of carbon. You may use any inorganic reagents necessary. Do not worry about arrow pushing (mechanism). (8 points) (Problem 15.11 in homework)

$$\frac{NBS}{Br} \frac{Br}{EH_2O} \frac{Mq}{EH_2O} \frac{MCPBA}{NH_3}$$

$$\frac{NBS}{Br} \frac{NaNH_2}{NH_3} \frac{MCPBA}{OH}$$

$$\frac{NH_3}{NH_3} \frac{MCPBA}{OH}$$

9) Show a synthesis of

Use only the reagents given as a source of carbon. You may use any inorganic reagents necessary. Do not worry about arrow pushing (mechanism). (8 points) (Problem 16.55 in homework)

$$C = C - H$$

$$\frac{\text{NaNH2}}{\text{HanH2}} + - C = C \cdot \frac{\text{CH}_3 - I}{\text{HanH2}} + - C = C - \text{CH3}$$

$$\frac{\text{HanH2}}{\text{Lindslar}} \rightarrow C = C - \text{CH3}$$

$$\frac{\text{Lindslar}}{\text{Catalyst}} \rightarrow C = C - \text{CH3}$$